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QUESTION. 12.1. Write
the structures of the
following compounds:

(i) α -Methoxypropional
dehyde (ii)

3-Hydroxybutanal (iii)

2-Hydroxycyclopentane
carbaldehyde

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TEXT QUESTIONS 8.1.

Silver atom has completely filled d orbitals (4d 10) in its ground state. How can you say that it is a transition element?

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Write structures of the following compounds:

- (i) 2-Chloro-3-methylpentane (ii) 1-Chloro-4-ethylcyclohexane

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QUESTIONS. 14.1.

Glucose or sucrose are soluble in water but cyclohexane and benzene (simple six membered ring compounds) are insoluble in water Explain. ...

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QUESTIONS SOLVED.

2.1. Calculate the mass
percentage of benzene
(C₆H₆) and carbon
tetrachloride (CCl₄) if
22 g of benzene is

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dissolved in 122 g of
carbon tetrachloride.

Ans: Mass of solution =
Mass of C₆H₆ + Mass
of CCl₄ = 22 g + 122
g = 144 g
Mass % of
benzene = $\frac{22}{144} \times$
 $100 = 15.28 \%$

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SOLVED. 11.1. Classify the following as primary, secondary and tertiary alcohols.

Ans: Primary alcohols: (i), (ii), (iii) Secondary alcohols: (iv), (v)

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