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It states that thermal energy can flow from colder objects to hotter objects only if work is done on the system.

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Third law of thermodynamics It states that absolute zero cannot be reached.

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...

16.1 Thermal Energy and Matter Heat flows spontaneously from hot objects to cold objects. • Heat is the transfer of thermal energy from one object

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to another because of a temperature difference.

Temperature is related to the average kinetic energy of the particles in an object due to their random motions through space.

Chapter 16 Thermal Energy and Heat

Physical Science

Chapter 16: Thermal Energy and Heat. Heat is the transfer of thermal energy from

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one object to another
as the result of a
difference in

Answers produces
heat.

Physical Science
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Energy and Heat ...

Chapter 16 Thermal
Energy and Heat
Section 16.1 Thermal
Energy and Matter
(pages 474–478) This
section defines heat
and describes how

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work, temperature, and thermal energy are related to heat. Thermal expansion and contraction of materials is discussed, and uses of a calorimeter are explained.

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Section 16.1

Thermal ...

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Section 161 Thermal

Energy and Matter

(pages 474-478) This

section defines heat

and describes how

work, temperature, and

thermal energy are

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related to heat

Thermal expansion and contraction of materials is discussed, and uses of a calorimeter are explained Reading Strategy (page 474)

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Chapter 16 Physics on Thermal energy - about convection, conduction and radiation as well as the use of insulation.

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Lynne_McCune.

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Temperature. Celcius.

temperature. the

transfer of thermal

energy from one object

to another beca.... the

measure of how hot or

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cold an object is compared to a refe....
Metric temperature scale on which water freezes at 0 degrees a....

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Summary 16.1 Thermal Energy and Matter
Heat flows spontaneously from hot objects to cold

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objects. • Heat is the transfer of thermal energy from one object to another because of a temperature difference.

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Section 161 Thermal Energy and Matter (pages 474–478) This section defines heat and describes how work, temperature, and thermal energy are related to heat Thermal expansion and contraction of materials is discussed, and uses of a calorimeter are explained Reading Strategy (page 474)

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Section 16.1 Thermal Energy and Matter

(pages 474–478) This section defines heat

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Thermal expansion and

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and uses of a

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explained Reading
Strategy (page 474)

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a temperature
difference.

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to the average kinetic
energy of the particles
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