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SECTION 1 Date CHAPTER 11 REVIEW
Gases Class SHORT ANSWER Answer the
following questions in the space
provided. b Pressure — orce For a
constant force, when the surface area is
tripled the surface area pressure is (a)

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doubled. as much. (c) tripled. 7-0 (d)
unchanged. Rank the following
pressures in increasing order. (c) 76 torr
(a) 50 kPa O, OOictbv-x

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SECTION 3 Date CHAPTER 11 REVIEW
Gases Class SHORT ANSWER Answer the
following questions in the space

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provided c c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol (b) divided by the mass of 1 mol nRT (c) multiplied by 224 L (d) divided by 224 L For the expression $V =$ (a) increasing P

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Answers

CHAPTER 11 REVIEW Gases SECTION 3

SHORT ANSWER Answer the following questions in the space provided

1 The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol (c) multiplied by 22.4 L (b) divided by the mass of 1 mol (d) divided by 22.4 L

2 For the expression $V = nRT/P$

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the properties of ideal gases. In this
chapter, you will study the predictions of
kinetic-molecular theory for gases in
more detail. This includes the
relationship among the temperature,

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pressure, volume, and amount of gas in a sample.

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Modern Chemistry 97 Gases CHAPTER
11 REVIEW Gases SECTION 3 SHORT
ANSWER Answer the following questions
in the space provided. 1. _____ The molar

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Answers

mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L. 2. _____ For the expression , P

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ANSWER . Answer the following 9 questions in the space provided. 1. c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 . L. (b) divided by the mass of 1 mol. (d) divided by 22.4 . L. 2. c For the expression $V = n \cdot T$,

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CHAPTER REVIEW Gases

CHAPTER 11 REVIEW Gases SECTION 2

SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances:

- increase a. temperature increases,
volume stays the same decrease b.

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Answers

volume increases, temperature stays the same

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(also atmospheric) SI unit of force.
definition & use of a barometer.
standard temperature & pressure (STP)
Definition of Dalton's law of partial pressures. Definitions & formulas for Boyle's, Charles', Gay-Lussac's, and combined gas laws

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Answers

CHAPTER 11 REVIEW. Gases. SECTION 3.
SHORT ANSWER Answer the following questions in the space provided. 1. The molar mass of a gas at STP is the density of that gas. (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L. 2.

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CHAPTER 11 REVIEW. Gases. SECTION 1.
SHORT ANSWER Answer the following
questions in the space provided. 1. _____
For a constant force, when the surface
area is tripled the pressure is.
(a)doubled. (b)a third as much.

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459. t's Monday morning, and Lilia is walking out of the chemistry building, thinking about the introductory lecture on gases that her instructor just presented. Dr.

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