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Rusting of iron and combustion of fuels are simply chemical reactions. Question 3.

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Section 11.1 Assessment. Describe the steps in writing a balanced chemical equation. Write the skeleton equation for

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the following reactions: Heating solid copper(II)sulfide in the presence of oxygen gas produces pure copper and sulfur dioxide gas. Iron metal and chlorine gas react to form solid iron(III)chloride.  $\text{CuS (s)} + \text{O}_2\text{(g)} \rightarrow \text{Cu (s)} + \text{SO}_2\text{(g)}$

## **Chapter 11: Chemical Reactions**



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or heat. reactant. ...

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equation. skeleton equation. catalyst. coefficient. A short, easy way to show a chemical reaction, using symbols. a chemical equation that does not indicate the relative amount.... a substance that initiates or accelerates a chemical reaction....

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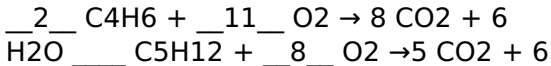
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## **Answer Key: Chemical Reactions Review**

As shown in Figure 11.3.1, applying a small amount of heat to a pile of orange ammonium dichromate powder results in a vigorous reaction known as the ammonium dichromate volcano. Heat, light, and gas are produced as a large

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pile of fluffy green chromium(III) oxide forms. We can describe this reaction with a chemical equation An expression that gives the identities and quantities of the ...

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Reactions Often charcoal briquettes provide the heat for barbeque grills through the burning of carbon. Have you ever felt the heat and smelled the smoke coming from a burning charcoal grill? The heat and smoke are the products of a combustion reaction. Combustion is one of the five general types of chemical

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Most chemical reactions can be classified into one or more of five basic types: acid-base reactions: A chemical reaction that has the general form  $AB + C \rightarrow AC + B$  or  $AB + CD \rightarrow AD + CB.$ , condensation reactions: A chemical reaction that has the general form  $A + B \rightarrow AB$ . Condensation reactions are the reverse of cleavage reactions.



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## **Chapter 11.6: Types of Chemical Reactions - Chemistry ...**

A chemical reaction can be concisely represented by a chemical 1. The substances that undergo a chemical change are the 2. The new substances formed in a chemical reaction are the 3. In accordance with the law of

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conservation of,4.

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